# Plant Design for a Production Process of Nanoasphalt Emulsion from Asbuton Rock 

Riny Yolandha Parapat Chemical Eng. Dept<br>National Institute of Technology (ITENAS) Bandung<br>Bandung, Indonesia<br>Email: rinyyolandha@itenas.ac.id

Jono Suhartono<br>Chemical Eng. Dept<br>National Institute of Technology<br>(ITENAS) Bandung<br>Bandung, Indonesia<br>Email: jonosuhartono@itenas.ac.id

Imam Aschuri<br>Civil Eng. Dept<br>National Institute of Technology<br>(ITENAS) Bandung<br>Bandung, Indonesia<br>Email: aschuri@itenas.ac.id

Michael Schwarze<br>Technische Chemie<br>Technische Universitat Berlin (TUB) Berlin, Germany<br>Email: Schwarze@tu-berlin.de

Reinhard Schomaecker<br>Technische Chemie<br>Technische Universitat Berlin (TUB)<br>Berlin, Germany<br>Email: Schomaecker@tu-berlin.de


#### Abstract

The largest deposit of natural asphalt in the world is located on Buton Island, Indonesia, which is around 677 million tons. With such abundance, Indonesia should be able to supply domestic asphalt needs, even has the potential to export abroad. However, the asphalt industries in Indonesia have not been able to process effectively the natural asphalt from Asbuton rock into a better quality of asphalt. Generally, they only focuses on resizing and separating the Asbuton rock according to the specifications of filler. This situation causes Indonesia has to import asphalt from abroad. Several studies and experiments have been done previously in lab-scale to produce high yield of nanoasphalt from Asbuton by combining microemulsion technique and Ultrasonication. Here we present a conceptual plant design of nanoasphalt production to process 121.8 tons/day of Asbuton rock as the raw material and produce 190 ton/day of nanoasphalt. The economic feasibility of processes of the Plant Design is analyzed. The economic evaluation in this work involves analyzing the capital and operating costs of the process to determine the return on investment (ROI), Break Even Point (BEP), Payback Period (PP), Internal Rate of Return (IRR). The result shows that the production of nanoasphalt is relative economical with ROI, BEP, PP and IRR are $31.54 \%, 24.5 \%, 2.47 \%, 35.46 \%$, respectively.


Keywords-Asbuton, nanoasphalt, microemulsion, ultrasonication

## I. Introduction

In recent years, Indonesian government has been stepping up development road infrastructure to build connectivity according to the Nawacita program and it needs the supply of asphalt about 1.3 million tons/year. On the other hand, Pertamina as the supplier of oil asphalt only be able to meet $30 \%$ of domestic asphalt needs. Even according to the estimation from Binamarga, it is likely that in year 2025, Pertamina will no longer be producing petroleum asphalt. Also, based on foreign trade data compiled by UN

Comtrade, it is recorded that Indonesia is one of the 10th largest asphalt importers in the world which worth of US \$ 371 million.

One way to overcome this, is to use natural asphalt from rocks found on the Buton Island and its surroundings as known as Asbuton. The amount of asphalt deposits on Buton Island are very large, that is 650 million tons. In fact, it is the greatest deposit in the world. With such abundance, Indonesia should be able to supply domestic asphalt needs, even has the potential to export abroad. However, currently, Indonesia is still unable to utilize effectively the abundance of the natural asphalt. This is indicated by the fact that Indonesia still has to import asphalt from neighboring countries. Singapore is the country of source for Indonesia's main asphalt. One of the State-Owned Enterprises (BUMN), PT Wijaya Karya Tbk (WIKA), is currently able to produce Asphalt Buton (Asbuton) but its production is still limited and relatively small compared to the amount imported.

In general, compare to petroleum asphalt, Asbuton has better characteristics such as: higher stability of marshall, higher dynamic stability of asphalt mix, better construction life, and more resistant to temperature changes. In addition, because Asbuton contains high aromatic and resinous ingredients, it has high adhesion and flexibility. Therefore, with those characteristics, Asbuton is suitable to be applied in high temperature locations (tropical) and for heavy loaded highway. Therefore, several researchers have attempted to extract asphalt from Asbuton. However, so far the process being implemented is still relatively expensive and involves hazardous solvents.

Nanotechnology has been increasingly intruded into the field of asphalt modification. Nanomaterial exhibits specific features novel properties compared to the bulk material due to its large surface area. Properties of nanomaterials can be
changed through controlling the size, regulating chemical composition, surface modification and controlling interactions between particles. Outstanding effects of nanomaterials are being brought to improve the performance of asphalt. Several nanomaterials used in asphalt modification have been studied by some researchers. Adding nanomaterials such as nanoclay, nanosilica, and nanotubes in asphalts will increase the viscosity of asphalt binders and improves the rutting and fatigue resistance of asphalt mixtures [1], [2]. Researchers anticipate nanotechnology to provide great potential in advancing asphalt pavement technology in the fields of materials design, manufacturing, properties, testing, monitoring and modeling [3], [4]. Specifically, focus areas in asphalt pavement analysis should include the bonds between aggregate, bonds between layers, properties of the mastic, self-repair and rejuvenation of binder, ageing (oxidation) effects and improvements in surface to tire properties [5], [6]. Nanoparticles for pavement materials is required to be non-hazardous low-cost products, due to them being spread over large volumes of material and being in almost direct contact with human activities. The reduction of energy requirements during construction of asphalt pavements through development of improved emulsions and reduction in binder viscosity at ambient temperatures through the introduction of micro-bubbles will not only lead to a potential energy cost saving, but also assist in the lowering of emissions during construction. The typical bitumen binder thickness coating around aggregate is in the order of a few microns [7], [8]. However, most studies on binder properties do not focus on this small dimension.

Nanoclays have very large aspect ratio with non-uniform size and shape. Adding $6 \%$ of nanoclays improves the permanent deformation or rutting behavior and enhance the resistance to aging of asphalt. Using nano calcium carbonate in asphalt mixtures can reduce the permanent deformation on asphalt pavement. The optimum performance is reached by adding $5 \%$ of these nanomaterials to the asphalt mixture [9], [10]. In Asbuton rock, there are naturally occurring minerals and subjected to natural variation in their formation. Various physical properties (such as stiffness and tensile strength, tensile modulus, flexural strength and modulus thermal stability) of the bitumen can be enhanced when it is modified with small amounts of nano-clay, on condition that the clay is dispersed at nano-scopic level. Generally, the elasticity of the nanoclay modified bitumen is much higher and the dissipation of mechanical energy much lower than in the case of unmodified bitumen [11], [12].

The mineral composition of Asbuton rock containing bitumen ( $\sim 30 \%$ ) found in Lawele are following: $\mathrm{CaCO}_{3}$ $72.90 \% \mathrm{SiO}_{2} 17.06 \%, \mathrm{Fe}_{2} \mathrm{O}_{3} 2.31 \%, \mathrm{Al}_{2} \mathrm{O}_{3} 1.94 \%, \mathrm{MgCO}_{3}$ $1.28 \%$ and the total other minerals about $5 \%[13]$. This shows that the mineral in Asbuton rock can be made as the nanoparticles. Instead of buying the expensive nanoparticles and added to the asphalt, the nanoparticles can be produced in-situ during the process. Moreover, adding nanoparticles to the very viscous material such as asphalt needs a very high energy and time to mix and make nanoparticles to be well dispersed in the asphalt mixture[1], [14], [15] . This method can anticipate all those disadvantages, because the nanoparticles are not added from outside but inside, thus they can be well mixed easily.

In this work, we present a Plant Design of nanoasphalt production from Asbuton rock based on our experimental
results in lab-scale. Several studies and experiments have been done previously in lab-scale to produce high yield of nanoasphalt from Asbuton by combining microemulsion technique and Ultrasonication.

## II. MATERIAL AND PROCESS DESCRIPTION

## A. Material

The raw materials used in the production of nanoasphalt include Asbuton rock, diesel oil as the solvent and sorbitan monooleate as the surfactant. The sorbitan monooleate surfactant required was produced in this plant, meanwhile the necessary Asbuton and diesel oil are obtained from other parties.

Asbuton contains high aromatic and resinous ingredients, so that asbuton has high adhesion and flexibility in the mixture. Another advantage is that the crack resistance due to weather and environmental consequences of asbuton is very high [16], [17]. The Asbuton needed is obtained from PT Wijaya Karya Bitumen which has a capacity of 66,000 tons/year at a price of IDR 270/ton.

Diesel oil (namely solar) is a fraction of clear brown yellow petroleum with a boiling point range between $250^{\circ} \mathrm{C}$ to $350^{\circ} \mathrm{C}$ which is also called the middle distillate. Diesel oil is chosen as a solvent in the nanoasphalt production process because of its abundant availability in Indonesia and its relatively cheap price. In addition, diesel fuel contains aromatic compounds which make diesel fuel very easy to extract asphalt solids. Diesel oil is also not reactive so that the extraction process avoids unexpected side reactions. The need for diesel for the nanoasphalt production process is obtained from PT Pertamina RU IV Cilacap with a capacity of 348,000 barrels per day at a price of IDR 9400/liter.

Surfactant is a substance that has the ability to reduce the surface tension of a medium and reduce the interfacial tension between two phases with different degrees of polarity. The term interface refers to the side between two phases which do not dissolve each other, while the term surface refers to the interface where one of the phases is air (gas). The surfactant structure can be described as a tadpole that has a head and tail. The head part of the surfactant is hydrophilic or polar and compatible with water, while the tail is hydrophobic or non-polar and is more attracted to oil/fat). The surfactant used in this nanoasphalt production process is sorbitan monooleate which has commercial name of Span 80. The raw materials for making Sorbitan monooleate include sorbitol, oleic acid, and both NaOH and $\mathrm{H}_{3} \mathrm{PO}_{3}$ solutions which function as catalysts.

Sorbitol $\left(\mathrm{C}_{6} \mathrm{H}_{14} \mathrm{O}_{6}\right)$ is a sugar alcohol which is a reduction product of glucose. In Indonesia, sorbitol is commonly produced from cassava root flour. In addition, sorbitol can also be found in red algae which contain $13.6 \%$ sorbitol [18]. This sorbitol was obtained from PT Sorini Argo Corporation Pasuruan which has a capacity of 120,000 tons/year at a price of IDR $16,200 / \mathrm{kg}$.

Oleic acid is a fatty acid found in animal and vegetable oils. Oleic acid is also called unsaturated fatty acids because it has double bonds in its C atomic chain. In its pure state, this acid is liquid with a light yellow or brownish color. The required oleic acid is obtained from PT Aneka Kimia Inti at a
price of IDR 13,700/kg. PT Aneka Kimia Inti has a capacity of 182,000 tons/ year.

Sodium hydroxide $(\mathrm{NaOH})$ also known as caustic soda is a type of caustic metal base. Sodium hydroxide is very soluble in water and will give off heat when dissolved. The required NaOH is obtained from PT Asahimas Chemical Cilegon at a price of IDR $4,500 / \mathrm{kg}$. PT Asahimas Chemical has a capacity of 700,000 tons/year.

Phosphoric acid or orthophosphoric acid is an inorganic mineral acid which has the chemical formula $\mathrm{H}_{3} \mathrm{PO}_{3}$. Phosphoric acid is a solid substance which at high temperatures will undergo dissociation. The $\mathrm{H}_{3} \mathrm{PO}_{3}$ required was obtained from PT Pradana Chemindo Perkasa at a price of IDR $13,700 / \mathrm{kg}$.

Tires are a composite material of isoprene natural rubber. Tires consist of rubber or polymer materials reinforced with synthetic fibers and steel which can produce a material with very high tensile strength, flexibility and high shear resistance. The used tire powder needed as an additive in the manufacture of surfactants was obtained from CV Gemilang RnP Jaya Makmur Bekasi at a price of IDR 2,700/kg.

## B. Process Description

The process of making nanoasphalt from Asbuton is carried out using a microemulsion extraction method with the help of solvents and surfactants. The solvent used is diesel oil and the surfactant used is sorbitan monooleate. The production process begins with the manufacture of surfactants and continues with the extraction of nanoasphalt as shown in the process flow diagram (PFD) in Fig. 1. The list of equipment in the PFD is given in Table 1. The process of making sorbitan monooleate surfactant is carried out in a packed bed reactor with a temperature of $170^{\circ} \mathrm{C}$ and a pressure of 1 atm . The process of making sorbitan monooleate is an esterification reaction using an acid-base catalyst, namely a mixture of NaOH and $\mathrm{H}_{3} \mathrm{PO}_{3}$. First, sorbitol powder, liquid oleic acid and NaOH crystals are mixed in the M-001 mixer at a temperature of $170^{\circ} \mathrm{C}$. On the other hand, mixing $\mathrm{H}_{3} \mathrm{PO}_{3}$ with water was carried out in the M-002 mixer which was also heated to $170^{\circ} \mathrm{C}$ to form a gasphase $85 \% \mathrm{H}_{3} \mathrm{PO}_{3}$ solution. A mixture of sorbitol, oleic acid and NaOH enters the top of reactor $\mathrm{R}-001$ while $\mathrm{H}_{3} \mathrm{PO}_{3}$ gas and water enters the bottom of reactor R-001. The reaction that occurs in R-001 is as follows.

$$
\mathrm{C}_{6} \mathrm{H}_{14} \mathrm{O}_{6}+\mathrm{C}_{18} \mathrm{H}_{34} \mathrm{O}_{2} \longrightarrow \mathrm{C}_{24} \mathrm{H}_{44} \mathrm{O}_{6}+2 \mathrm{H}_{2} \mathrm{O}
$$

The gas-phase reaction product, namely water and $\mathrm{H}_{3} \mathrm{PO}_{3}$ then exits through the top of the reactor, then enters C-001 to be condensed. The $\mathrm{H}_{3} \mathrm{PO}_{3}$ gas and water which has been condensed are then cooled to a temperature of $30^{\circ} \mathrm{C}$ and enter the storage tank. The results of the search phase reaction include the sorbitan monooleate product, the unreacted reactants, namely sorbitol and oleic acid, and the NaOH catalyst then pumped by P-005 to the distillation column D001 to be separated.

In the distillation column $\mathrm{D}-001$, the reaction product is heated at a pressure of 1 atm until the boiling point of oleic acid is $360{ }^{\circ} \mathrm{C}$. This causes the oleic acid and sorbitol to evaporate into a distillate while the sorbitan monooleate and NaOH products are left as the bottom. The resulting distillate then passes through C-002 to be condensed and cooled before entering the storage tank. Meanwhile, the bottom consisting of the sorbitan monooleate tie and NaOH is
pumped into the microemulsion extraction process that occurs in the Sonifier SF-001.

Sonifier SF-001 works at a pressure of 1 atm and a temperature of $60^{\circ} \mathrm{C}$. The bait that enters sonifier SF-001 is asbuton rock which is crushed to $200 \mu \mathrm{~m}$ using Ball Mill BM-001. The crushed Asbuton rock then enters screen S-001 to be filtered from rocks that are still large in size. Large rocks are returned to $\mathrm{BM}-001$ via belt conveyor $\mathrm{BC}-005$, while rocks that pass the screening then enter the sonifier. Apart from Asbuton rock and sorbitan monooleate, solvent is simultaneously pumped to the sonifier. The nanoasphalt production process also uses a 40 mesh rubber additive.

The result of the sonification process on SF-001 in the form of a slurry is then pumped by P- 010 to the M-003 mixer. The mixture in the M-003 mixer is heated to $150^{\circ} \mathrm{C}$ where the rubber additive is melted so that it is mixed more homogeneously. From the M-003 mixer, the extracted product is then pumped into the SP-001 separator and allowed to stand for some time until two layers are formed. The top layer is the nanoasphalt that is formed while the lower layer is the residue of Asbuton rock. The nanoasphalt product or top layer is cooled by cooler C-004 to a temperature of $30^{\circ} \mathrm{C}$ and stored in storage tank T-012. Meanwhile, the lower layer, namely the Asbuton rock residue, still contains quite a lot of bitumen and minerals, so it is washed in two stages. This washing is done by adding diesel solvent to the residue in the M-004 mixer (first stage). This mixer works at a pressure of 1 atm and a temperature of $150^{\circ} \mathrm{C}$. The washing product from the mixer then enters the separator SP-002 for separation as before until two phases are formed. The top layer in the form of nanoasphalt is then pumped into the distillation column D-002 while the lower layer residue enters the M-005 mixer and SP-003 separator for the second washing process. The top layer obtained from the two washing processes then goes into the D-002 distillation column to be separated between the nanoasphalt and solar solvent. The distillation column D-002 operates at a pressure of 1 atm and a temperature of $250^{\circ} \mathrm{C}$, which is at the boiling point of diesel so that the diesel is evaporated into a distillate and the nanoasphalt product is left as the bottom. The distillate is then condensed and cooled before entering the storage tank.

The washed nanoasphalt product will certainly have less nanoparticle composition when compared to the main nanoasphalt product. Therefore, the washed nanoasphalt product that comes out of D-002 then enters the M-006 mixer at a pressure of 1 atm and a temperature of $150^{\circ} \mathrm{C}$. In this mixer, the washed nanoasphalt product is mixed with the Asbuton rock residue which still has remaining mineral content. The Asbuton rock residue is of course crushed first to nano-sized at BM-002. The mixing between the residue and the nanoasphalt will make the washed nanoasphalt product have a nanoparticle composition that is equivalent to the main nanoasphalt product.

## III. ECONOMIC EVALUATION

The calculation in economic evaluation is based on the following assumptions:

- the bank interest rate of $12 \%$
- the CE-Index 2020 is 597.05
- the tax applied is $25 \%$ on gross profit
- the loan is $10 \%$ from the total capital investment (TCI)
- the plant's life is 20 years


## A. Total Capital Investment (TCI)

Capital investment is required for any industrial process, therefore classifying the required investment is an important part of a plant design project. The total investment for a process consists of physical equipment, plant equipment and facilities, as well as the working capital required to pay salaries, store raw materials and products, and manage other special items [19].

Total Capital Investment (TCI) is the sum of the Total Plant Cost (TPC) and Other Capital Requirements (OCR).

| T-009 | Solar Tank | R-001 | Reactor |
| :---: | :---: | :---: | :---: |
| T-010 | Landfill Asbuton | BM-001 | Ball Mill 1 |
| T-011 | Rubber Silo | BM-002 | Ball Mill 2 |
| T-012 | Nanoasphalt I Tank | D-001 | Distillation Column 1 |
| T-013 | Distillate II Tank | D-002 | Distillation Column 2 |
| T-014 | Nanoasphalt II Tank | SP-001 | Separator 1 |
| T-015 | Residue Silo | SP-002 | Separator 2 |
| M-001 | Mixer 1 | SP-003 | Separator 3 |
| M-002 | Mixer 2 | H-003 | Heater 3 |
| M-003 | Mixer 3 | H-004 | Heater 4 |
| M-004 | Mixer 4 | H-005 | Heater 5 |
| M-005 | Mixer 5 | RB-001 | Reboiler 1 |
| M-006 | Mixer 6 | RB-002 | Reboler 2 | The Total Plant Cost (TPC) is presented in the Table II.



Fig. 1. Process Flow Diagram of Nanoasphalt Prodcution

TABLE I. EQuipment in The Plant

| Unit <br> Code | Unit Operation | Unit <br> Code | Unit Operation |
| :---: | :---: | :---: | :---: |
| T-001 | Oleic Acid Tank | C-001 | Cooler 1 |
| T-002 | Sorbitol Silo | $\mathrm{C}-002$ | Cooler 2 |
| T-003 | NaOH Silo | $\mathrm{C}-003$ | Cooler 3 |
| T-004 | Water Tank | $\mathrm{C}-004$ | Cooler 4 |
| T-005 | $\mathrm{H}_{3} \mathrm{PO}_{3}$ Silo | $\mathrm{C}-005$ | Cooler 5 |
| T-006 | $\mathrm{H}_{3} \mathrm{PO}_{3}$ Tank | $\mathrm{C}-006$ | Cooler 6 |
| T-007 | Distillate I Tank | $\mathrm{H}-001$ | Heater 1 |
| T-008 | SPAN 80 Tank | $\mathrm{H}-002$ | Heater 2 |

The total cost for OCR is Rp. 580.626.482.799, which comprises of:

- Off-site facilities such as the cost of packaging, warehouse, transportation and offices which is $0.15^{*}$ TPC $=$ IDR 193.542.160.933
- Plant Start-up which is $0.1 * T P C=I D R 129.028 .107 .288$
- Working capital such as the cost of raw material, salaries and others during the factory operation which is $0.2 *$ TPC $=$ IDR 258.056.214.577

Therefore, to produce nanoasphalt of 68000 tons/year and sorbitan monooleate of 5000 tons/year, PT Hay Asphalt requires investment costs in the form of Total Capital Investment (TCI) of IDR 1.870,907,555,6877.

## A. Production Cost Estimation

Production cost estimation is divided into calculation of total sales, itemized cost, labor related cost, capital related cost and sales related cost, manufacturing cost and interest. Sales calculation is made by multiplying the number of products produced per year with the estimated selling price of the product on the market. Total sales of the plant is presented in Table III.

TABLE II. PLANT Cost Estimation

| Component | Value (IDR) | Assumption |
| :---: | :---: | :---: |
| Purchased Equipment | 256,772,352,813 | The costs of all the equipment include process equipment and auxiliary equipment |
| Piping | 109,128,249,945 | The plant process mostly involves solid and liquid, therefore it does not need too much piping |
| Electrical | 25,667,235,281 | All of the plant's electricity needs are supplied by PLTU. |
| Instrumentation | 89,870,323,484 | All the unit process are equipped with an instrumentation system and this unit is managed as well as possible so that the production process is more effective. |
| Utilities | 134,805,485,227 | The utilities that required in the plant are steam, water and compressed air. |
| Insulation | 20,541,788,225 | The plant involves thermal energy in many process equipment where this energy must not be wasted so that it can be utilized. |
| Painting, <br> Fireproofing, <br> Safety | 25,677,235,281 | Plant safety and robustness must be properly maintained so that no undesirable things happen. |
| Yard Improvements | 12,838,617,640 | The plant will be built on the green land in industrial areas, therefore the construction of sidewalks, parks, drainage, etc. does not require large costs. |
| Environmental | 51,354,470,562 | The plant produced only a small amount of waste, therefore the waste treatment costs are not too large. |
| Buildings | 134,805,485,227 | The production of sorbitan monooleate is carried out indoors to maintain the purity and minimize the contamination of the product. |
| Land | 12,836,617,640 | The plant will be built in an industrial area so that land costs are not too large. |
| Construction, Engineering | 192,579,264,610 | Good construction and techniques are used to minimize problems during the production process. |
| Contractors fee | 70,612,397,023 | The plant construction are handled by the local contractors (Indonesia) so the fee is not too costly. |
| Contingency | 21,966,667,586 | The nanoasphalt production process is a process that is still |


|  |  | rarely encountered so it <br> requires a large amount of <br> unexpected funds. |
| :--- | :--- | :--- |
| Total Plant <br> Cost (TPC) | IDR $\mathbf{1 , 2 9 0}, \mathbf{2 8 1 , 0 7 2 , 8 8 7}$ |  |

The production cost is the cost required by the plant during the production process. The production costs required consists of variable cost and fixed cost

Variable costs are costs required by the plant which depend on the amount of plant production capacity and other costs that support the production process. Variable cost is composed of itemized cost, labor related cost, capital related cost, and sales related cost, as presented in Table IV.

TABLE III.
Total Sales

| Product | Production <br> (ton/year) | Price <br> (IDR/ton) | Total Price <br> (IDR/year) |
| :--- | :---: | :---: | :--- |
| Nanoasphalt <br> (high quality) | 64680 | $23,920,224$ | $1,547,149,049,806$ |
| Nanoasphalt <br> (grade B) | 2950 | $23,000,000$ | $67,856,949,864$ |
| Sorbitan <br> monooleate | 2074 | $88,142,080$ | $182,844,530,286$ |
| Residue <br> (nanoparticle) | 2401 | $4,131,660$ | $9,921,990,589$ |
| Total (IDR) |  |  |  |$| \mathbf{1 , 8 0 7 , 7 7 2 , 5 2 0 , 5 4 6}$.

Itemized cost calculations include the calculation of the cost of purchasing raw materials, utilities and operating labor salaries. The raw material includes Asbuton, sorbitol, oleic acid, $\mathrm{H}_{3} \mathrm{PO}_{3}, \mathrm{NaOH}$, and rubber crumbs, whereas the utility comprises of electricity and boiler fuel. The operating labor costs is obtained from the number of operating employees multiplied by the annual salary. The costs of other components in Table IV is calculated based on the factors whose values are estimated based on the rule of thumb [20].

| TABLE IV. | VARIABLE COST |
| :--- | :---: |
| Variable Cost Cost (IDR) <br> b. Labized Cost  <br> Raw Material $378,340,366,312$ <br> Utility $13,650,450,853$ <br> Operating Labor $12,000,000,000$ <br> Cost  <br> Payroll Overhead $2,640,000,000$ <br> Supervisory, Misc. Labor $2,400,000,000$ <br> Laboratory Charge $1,800,000,000$ <br> Capital Related Cost  <br> Maintenance $96,771,080,466$ <br> Operating Supplies $28,224,896,469$ <br> Environmental $44,353,411,880$ <br> dales Related Cost  <br> Packaging, storage $63,272,038,219$ <br> Distribution and sales $108,466,351,232$ <br> R\&D $9,038,862,602$ <br> Total Variable Cost IDR $749,179,592,743$ |  |

Fixed costs are costs that are incurred regularly every time which not depend on the plant production capacity. Fixed costs are presented in the Table V below.

TABLE V. Fixed Cost

| Variable Cost | Cost (IDR) |
| :--- | :---: |
| aabor Related Cost |  |
| b. Capital Related Cost |  |
| Payroll Overhead | $2,640,000,000$ |
| c. Sales Related Cost |  |
| Local Taxes, Insurance | $64,514,053,644$ |
| Plant Overhead Cost | $48,385,540,233$ |
| Patent \& Royalties |  |

In the calculation of production cost, there is a component called depreciation. Depreciation is a systematic depreciable allocation cost of an asset over its useful life. These costs affect the financial statements, including the company's net profits. There are three methods to determine depreciation, namely straight line, double declining balance and sum of year digits. It is assumed that the plant life is 20 years without salvage value, and the land cost is IDR $12,838,617,640$. The calculation result for depreciation value using the straight line method in 20 years is IDR $63,872,122,762 /$ year.

## IV. Feasibility Analysis

Project feasibility analysis is useful for avoiding the worst prospect of the plant. The following method used for this analysis are Return on Investment (ROI), Break Even Point (BEP), Payback Period (PP), Internal Rate of Return (IRR), and Present Value (NPV). The analysis method is calculated from the cash flow data during the plant's establishment.

## A. Return on Investment (ROI)

Return on Investment (ROI) is a profitability ratio that measures the efficiency of an investment by comparing net income to the total cost or capital invested. ROI shows the rate of return on capital or investment per year. The ROI value, which is getting bigger each year, shows that the plant is more feasible to build (Table VI). The plant is considered to be feasible if the ROI > MARR. The MARR (Minimum Acceptable Rate of Return) is calculated by the net-profit average divided by TCI, which is equal to $20.75 \%$. Because ROI > MARR, the project is feasible.

TABLE VI.
Return on Investment (ROI)

| Year | ROI before Tax | ROI after Tax |
| :---: | :---: | :---: |
| 1 | $11,28 \%$ | $15,05 \%$ |
| 2 | $14,62 \%$ | $19,49 \%$ |
| $3-20$ | $21,24 \%$ | $28,31 \%$ |

## B. Break Even Point (BEP)

Break Even Point (BEP) is a point where the resulting production capacity can cover all production costs without any profit or loss or Cash Flow is zero. The smaller the BEP value indicates the more feasible the factory is to build. The BEP value of the plant before depreciation and after depreciation is shown in Table VII below.

TABLE VII. Break Even Pont (BEP)

| Year | BEP before Depreciation | BEP after Depreciation |
| :---: | :---: | :---: |
| 1 | $45,43 \%$ | $34,30 \%$ |
| 2 | $39,75 \%$ | $30,01 \%$ |
| $3-20$ | $31,80 \%$ | $24,01 \%$ |

## C. Payback Periode (PP)

Payback period is time required after startup to recover the total capital investment. Here, the payback period since the plant started operating to pay off the initial investment from the earned revenue is calculated by dividing Total Cost Investment (TCI) with the cash flow. The MARR (Minimum Acceptable Rate of Return) is calculated by the net-profit average divided by TCI, which is equal to $20.75 \%$. The PP acceptable which is calculated by the Total Plant Cost/((MARR * TCI) + Depreciation), which is equal to 3.5 years.

## D. Internal Rate of Return (IRR)

Internal Rate of Return (IRR) is the highest discount rate by calculating the nominal discounted rate (interest rate) which equates the present value of an investment with the present value of future net cash receipts at the time of the NPV value $=0$. The plant is considered to be feasible if IRR> deposit interest rate. This plant has an IRR value of $37.04 \%$ which is higher than the deposit interest rate ( $7.31 \%$ )

## E. Present Value (PV)

A plant can be considered to be feasible if the Present Value (NPV) is greater than zero at the end of the plant's life. The income from the plant after discounted (minus IRR) is presented in Table VI. Because the PV $>0$ at the end of the plant's life, then the plant is considered to be feasible. Fig. 2 shows that the depreciation allowance ceases in 7.5 years.


Fig. 2. Cummulative Cash Flow Diagram for the evaluation of the project

## F. Sensitivity Analysis

Sensitivity analysis is a method that aims to determine the level of sensitivity of the plant's feasibility to changes in
factors that are considered in economic analysis. These factors include sales, production costs, raw materials, and utilities. The sensitivity analysis is depicted as a spider web curve in Fig. 3.

As can be seen from Fig. 3, the four factors that influence the profits earned by the plant. If sales increase, the profits will also increase. Meanwhile, if raw materials, utilities and production costs increase, then the profits will decrease. The Sales Curve has a sharp increase which makes this factor have a big influence on the continuity of the plant if there is an increase or decrease. In addition, the three other factors, namely raw materials, costs and utility, have a sloping curve, even very gentle for the factor of complexity. This means that these three factors have a lower influence than the sales factor.


Fig. 3. Sensitivity analysis which shows the change in the NPV of the project cash flow at 20 years of production

However, if these three factors experience an increase that occurs simultaneously or continuously, then the factory will also experience losses. In other words, it can be said in total that the plant can be established.

## V. CONCLUSION

The plant economic analysis determines whether a plant is feasible to be established or not. The analysis result shows that the plant is commendable to be founded, because it fulfils the feasibility criteria, such as:

1. ROI ( $21.24 \%$ ) > MARR ( $20.75 \%$ )
2. PP ( 3.50 years) $<$ PPacceptable ( 3.57 years)
3. IRR $(28.97 \%)>$ Interest Rate ( $12 \%$ )
4. PV $($ IDR $2,931,836,681)>0$ at the end of factory life

The plant's sensitivity shows that if the sales value has decreased by $30 \%$, the factory will experience a loss. However, the plant economic is sensitive to utility and raw material is also insensitive to changes in raw material price increases and also prices for utilities, this explains that this factory has a low sensitivity level. With the establishment of this plant, it will certainly have a good and beneficial impact on Indonesia because it can reduce the need for asphalt and
surfactant imports, increase employment, improve the welfare of the surrounding community, and also increase the country's foreign exchange by having the opportunity to export abroad.

## AcknowLedgment

The Authors are very grateful to the Ministry of Research and Technology/National Research and Innovation of Indonesia (Kemenristekdikti/BRIN) for the financial support. Great support from LPPM-Itenas is also acknowledged. Thanks to the former students namely Ali Akbar, Abdul Mazid Yusuf, and Hani Tania for working out the plant design. Great appreciation also goes to our colleagues in Chemical Engineering Department-ITENAS Bandung, and in TC-08 group at TU-Berlin, Germany for their support.

## References

[1] J. Yang and S. Tighe, "A Review of Advances of Nanotechnology in Asphalt Mixtures," Procedia - Social and Behavioral Sciences, vol. 96, no. Supplement C, pp. 1269-1276, Nov. 2013
[2] H. Yao, Q. Dai, Z. You, M. Ye, and Y. K. Yap, "Rheological properties, low-temperature cracking resistance, and optical performance of exfoliated graphite nanoplatelets modified asphalt binder," Construction and Building Materials, vol. 113, no. Supplement C, pp. 988-996, Jun. 2016.
[3] P. Calandra, V. Loise, M. Porto, C. Oliviero Rossi, D. Lombardo, and P. Caputo, "Exploiting Nanoparticles to Improve the Properties of Bitumens and Asphalts: At What Extent Is It Really Worth It?," Applied Sciences, vol. 10, no. 15, Art. no. 15, Jan. 2020.
[4] H. Ezzat, S. El-Badawy, A. Gabr, E.-S. I. Zaki, and T. Breakah, "Evaluation of Asphalt Binders Modified with Nanoclay and Nanosilica," Procedia Engineering, vol. 143, no. Supplement C, pp. 1260-1267, Jan. 2016.
[5] Gh. Shafabakhsh, S. M. Mirabdolazimi, and M. Sadeghnejad, "Evaluation the effect of nano-TiO2 on the rutting and fatigue behavior of asphalt mixtures," Construction and Building Materials, vol. 54, pp. 566-571, Mar. 2014.
[6] L. Han, M. Zheng, J. Li, Y. Li, Y. Zhu, and Q. Ma, "Effect of nano silica and pretreated rubber on the properties of terminal blend crumb rubber modified asphalt," Construction and Building Materials, vol. 157, pp. 277-291, Dec. 2017.
[7] H. A. Omar, N. I. Md. Yusoff, M. Mubaraki, and H. Ceylan, "Effects of moisture damage on asphalt mixtures," Journal of Traffic and Transportation Engineering (English Edition), vol. 7, no. 5, pp. 600628, Oct. 2020.
[8] R. S. Mullapudi, S. L. A. Noojilla, and S. R. Kusam, "Effect of initial damage on healing characteristics of bituminous mixtures containing reclaimed asphalt material (RAP)," Construction and Building Materials, vol. 262, p. 120808, Nov. 2020.
[9] S. Bagshaw, T. Kemmitt, M. Waterland, and S. Brooke, "Effect of blending conditions on nano-clay bitumen nanocomposite properties," Road Materials and Pavement Design, pp. 1-22, May 2018.
[10] H. Yao et al., "Evaluation of Asphalt Blended With Low Percentage of Carbon Micro-Fiber and Nanoclay," JOURNAL OF TESTING AND EVALUATION, vol. 41, pp. 278-288, Mar. 2013.
[11] R. Muniandy, R. Bt, R. Yunus, H. Salihudin, and E. Aburkaba, "Effect of Organic Montmorillonite Nanoclay Concentration on the Physical and Rheological Properties of Asphalt Binder," Australian Journal of Basic and Applied Sciences, vol. 7, pp. 429-437, Jan. 2013.
[12] M. Sivakumar and M. V. L. R. Anjaneyulu, "Fatigue Characteristics of Nano-clay Modified Bituminous Concrete," Transportation Research Procedia, vol. 17, pp. 124-133, Dec. 2016.
[13] A. A., M. Nawawi, N. Usman, and M. Hamzah, "A preliminary mineralogical evaluation study of natural asphalt rock characterization, southeast Sulawesi, Indonesia," Arabian Journal of Geosciences, vol. 9, Apr. 2016.
[14] A. Mohajerani et al., "Nanoparticles in Construction Materials and Other Applications, and Implications of Nanoparticle Use," Materials (Basel), vol. 12, no. 19, Sep. 2019.
[15] J. Su, "Chapter 6 - Role of nanoparticles in self-healing of polymeric systems," in Self-Healing Polymer-Based Systems, S. Thomas and A. Surendran, Eds. Elsevier, 2020, pp. 141-165.
[16] M. Nasikin, I. M. Devianto, and B. Susanto, "ASPHALT PRODUCTION FROM ASBUTON BY SONICATED EXTRACTION OF CARBONATE SOLIDS USING ACIDIC BRINE WATER SOLUTION," undefined, 2013.
[17] I. M. Devianto, llyin A. Budianta, B. H. Susanto, and M. Nasikin, "Asphalt Production from Asbuton Rock by Extraction Using Weak Acid," Chemical and Materials Engineering, vol. 1, no. 2, pp. 3542, Aug. 2013.
[18] I. Goldberg, Functional Foods: Designer Foods, Pharmafoods, Nutraceuticals. Springer Science \& Business Media, 2012.
[19] Peter, M. Timmerhaus, K. Plant Design and Economics for Chemical Engineers, 5th ed.; McGraw-Hill: New York,NY, USA, 2003
[20] D. E. Garrett, Chemical Engineering Economics. Springer Netherlands, 1989

